

NOVE FIZIČKOHEMIJSKE METODE

TEORIJSKA ANALIZA
ULTRABRZIH PROCESA

MILENA PETKOVIĆ

- ultrabrzi procesi
- vremenski nezavisna Šredingerova jednačina
- vremenski zavisna Šredingerova jednačina

ULTRABRZI PROCESI

Ultrabrzim procesima nazivamo procese koji se odvijaju u opsegu od nekoliko atosekundi ($10^{-18}s$) do nekoliko nanosekundi ($10^{-9}s$).

- hemijske reakcije inicirane sudarom čestica ili povišenjem temperature
- interakcija čestica sa elektromagnetskim zračenjem: fotofizika i fotohemija

OSNOVNI ALAT TEORETIČARA

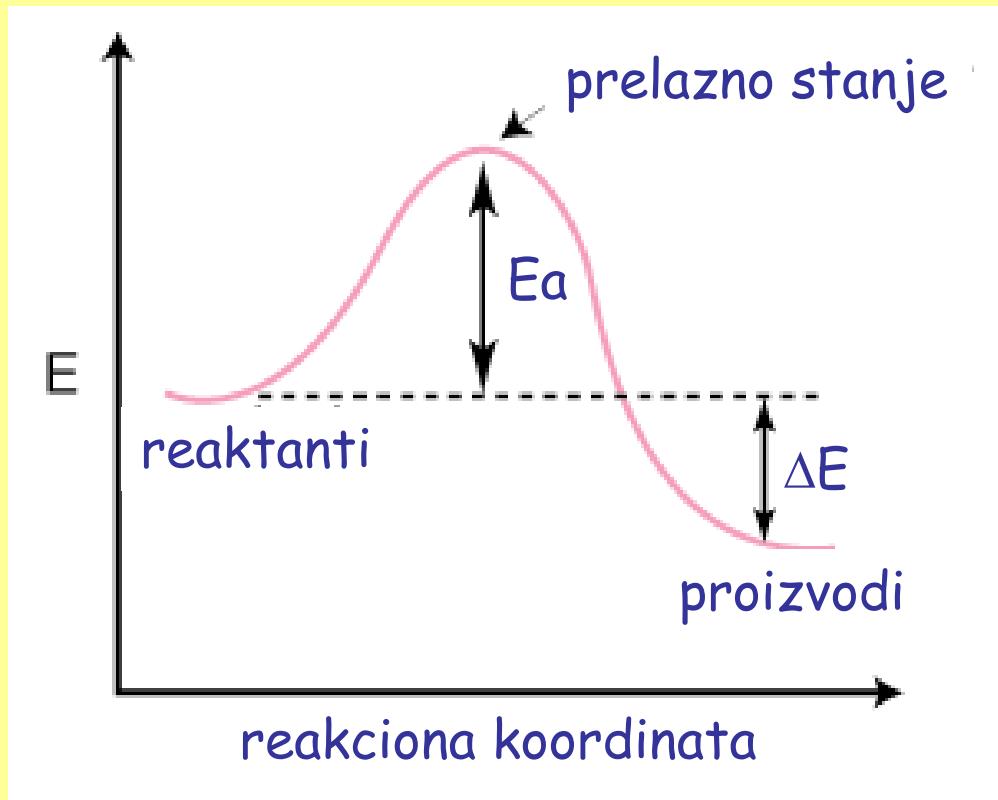
vremenski nezavisna Šredingerova jednačina

$$\hat{H}\Psi = E\Psi$$

vremenski zavisna Šredingerova jednačina

$$i\hbar \frac{\partial \Psi(t)}{\partial t} = \hat{H}(t)\Psi(t)$$

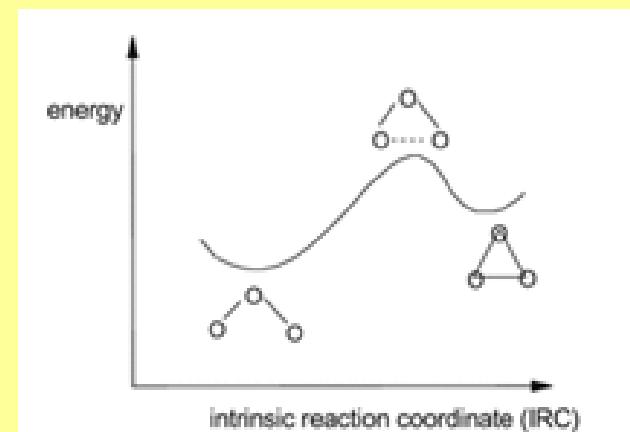
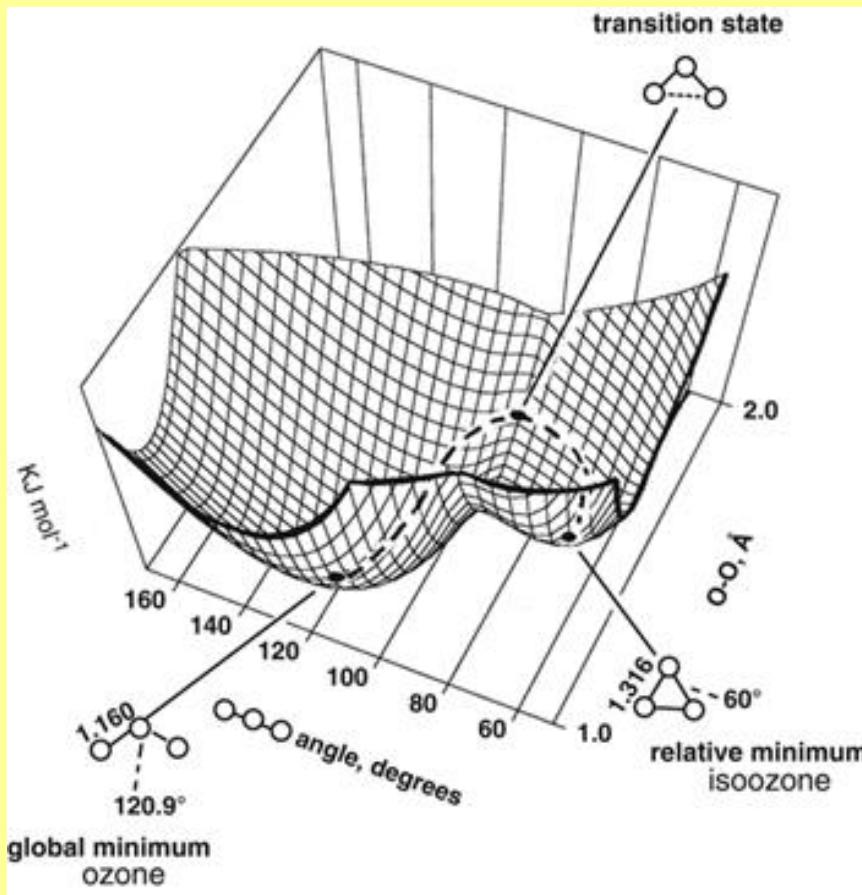
PRELAZNO STANJE



vreme života prelaznog stanja $\approx 10 - 100 \text{ fs}$

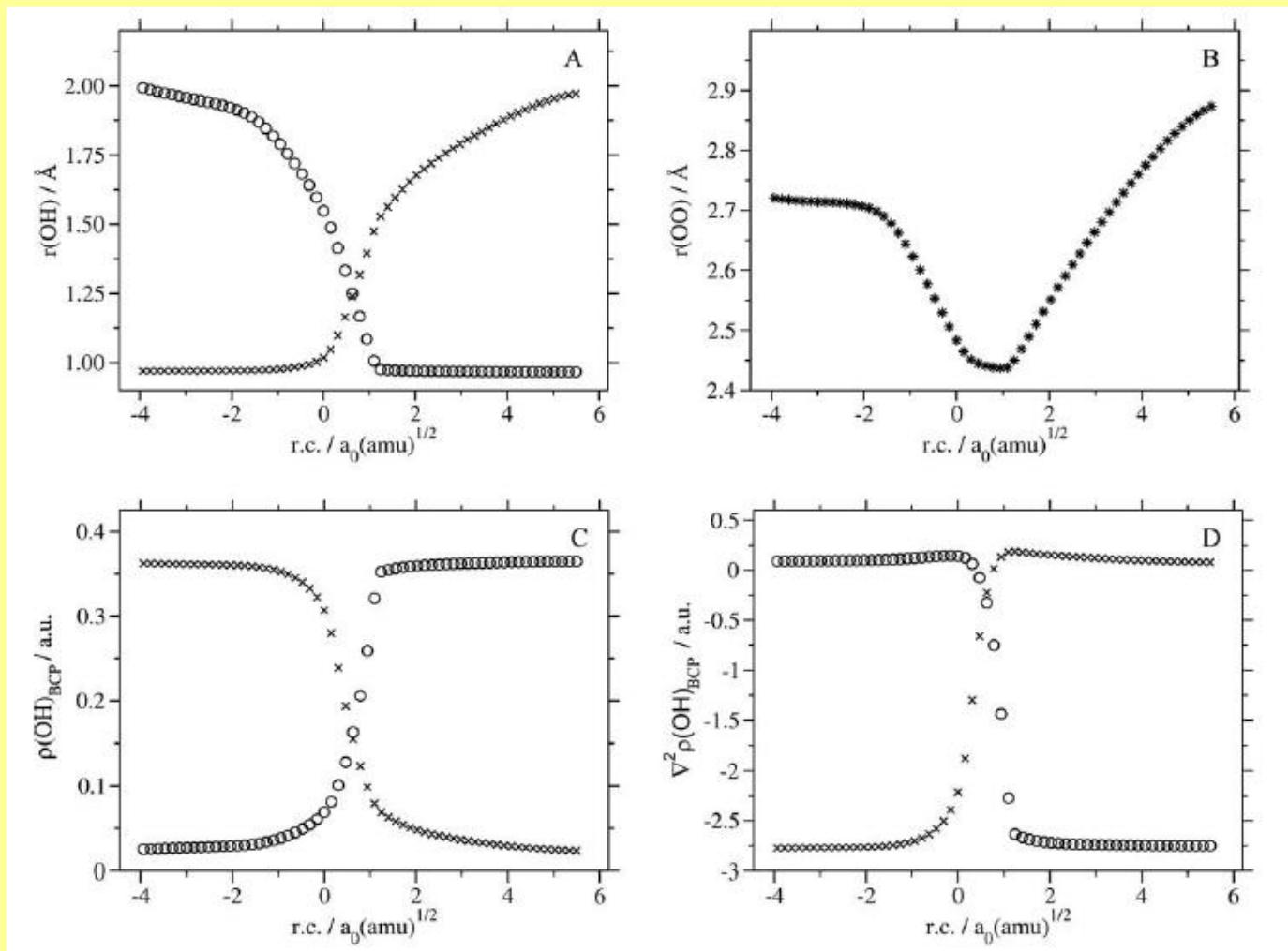
- ultrabrzi procesi
- vremenski nezavisna Šredingerova jednačina
- vremenski zavisna Šredingerova jednačina

PROMENE DUŽ PUTA MINIMALNE ENERGIJE

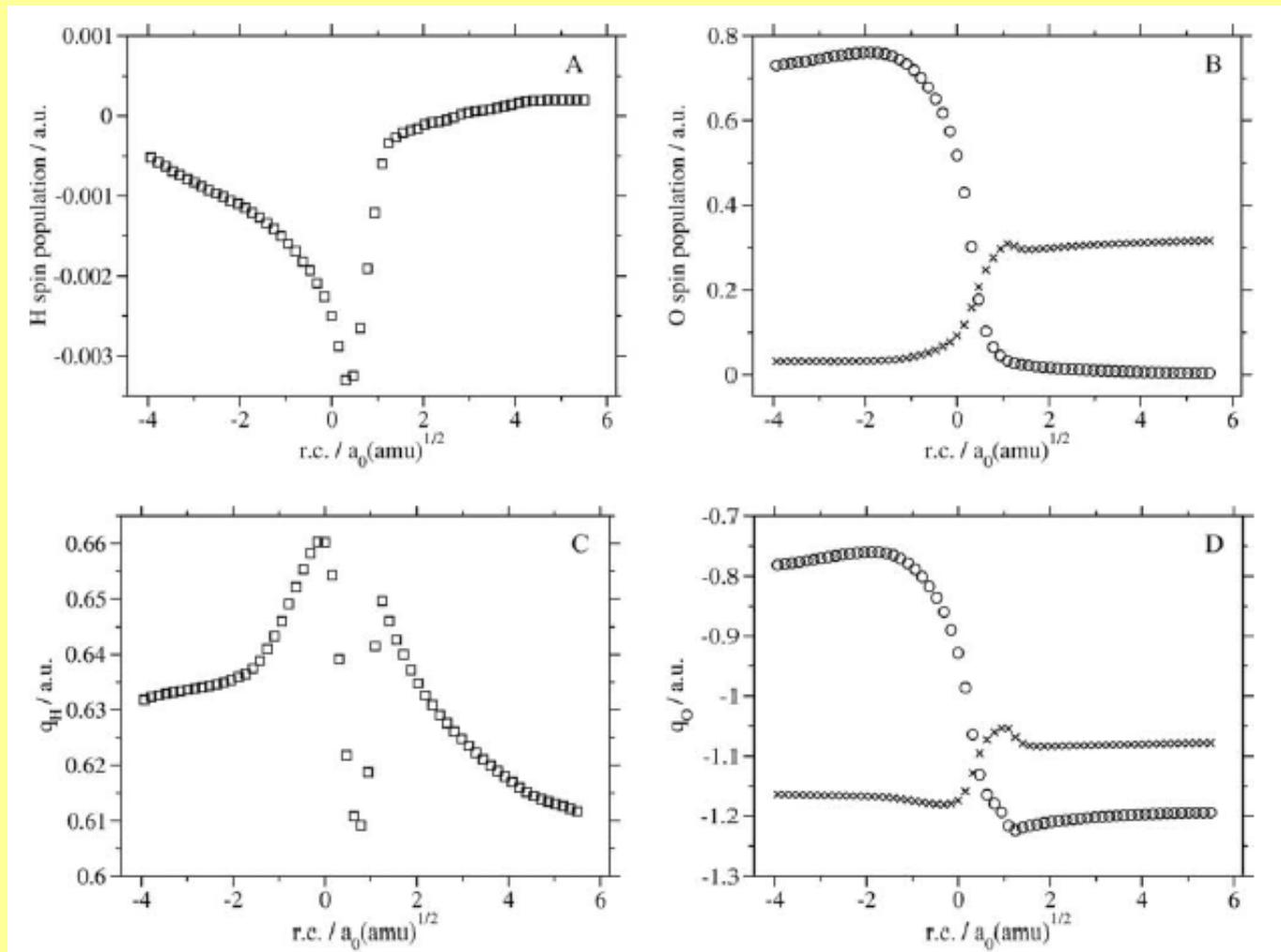
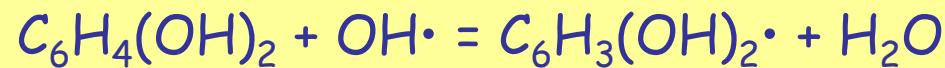


E. G. Lewars, The concept of the Potential Energy Surface in Computational Chemistry (Springer, Cham, 2016)

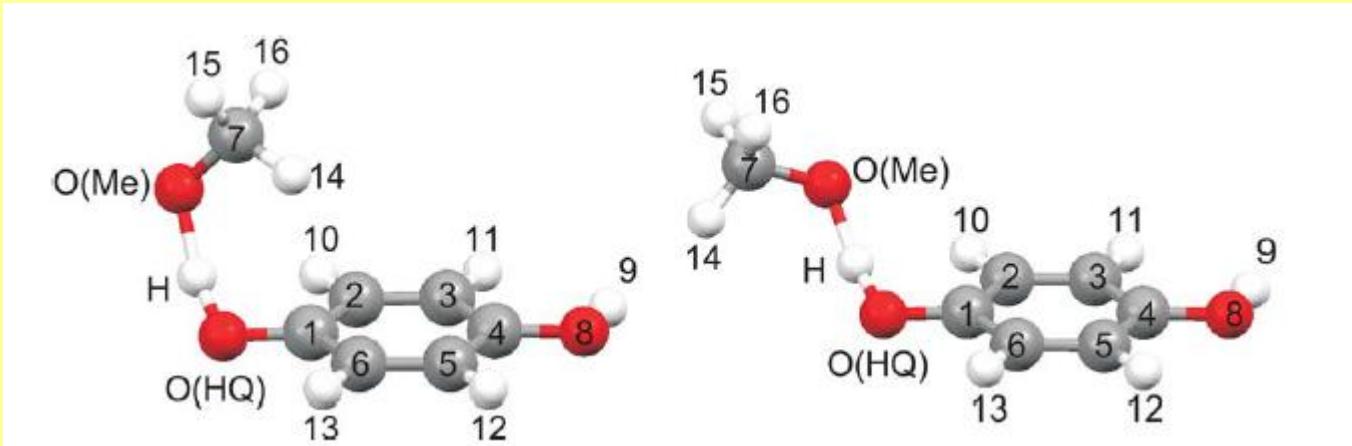
PROMENE DUŽ PUTA MIN. ENERGIJE (1)



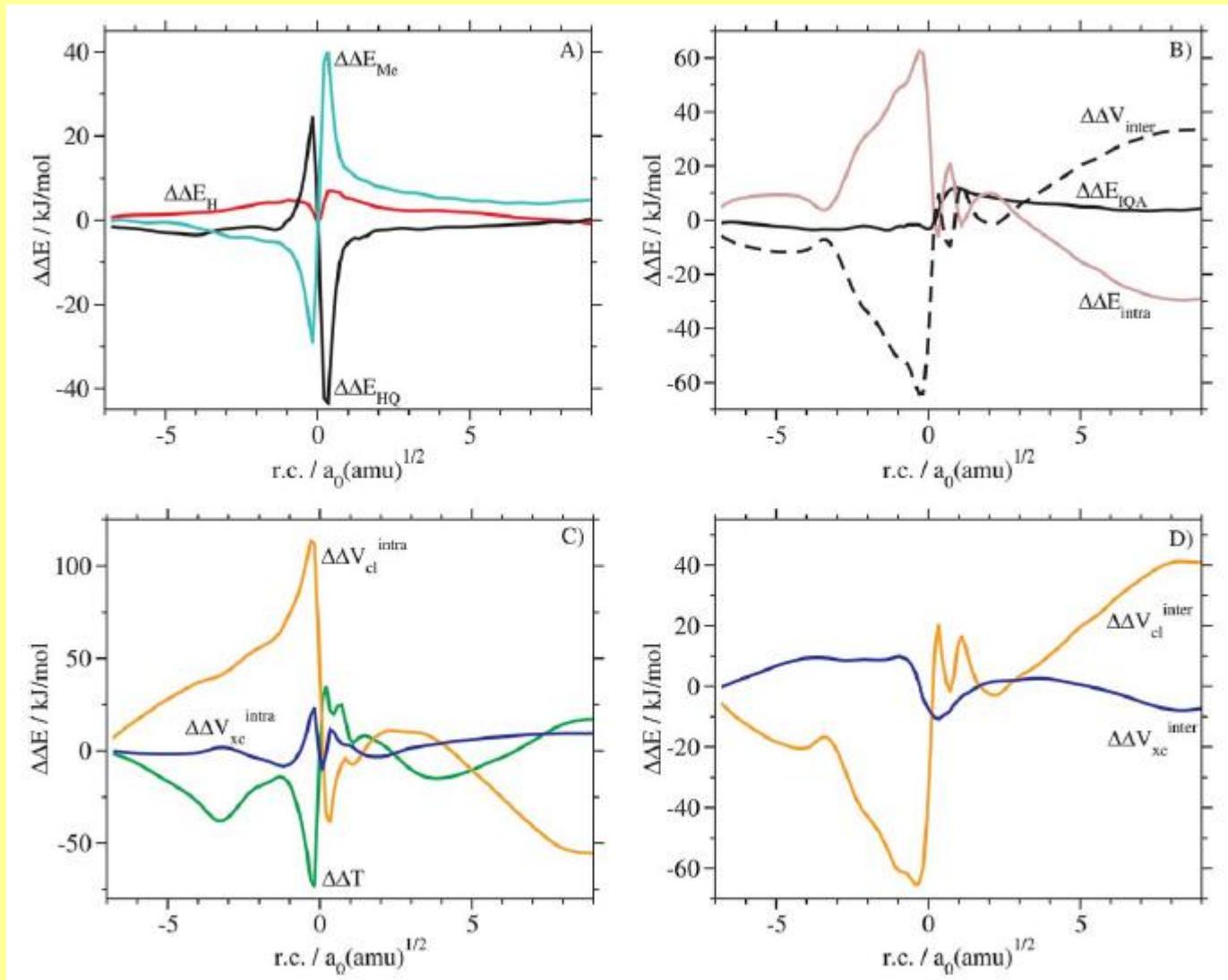
PROMENE DUŽ PUTA MIN. ENERGIJE (2)



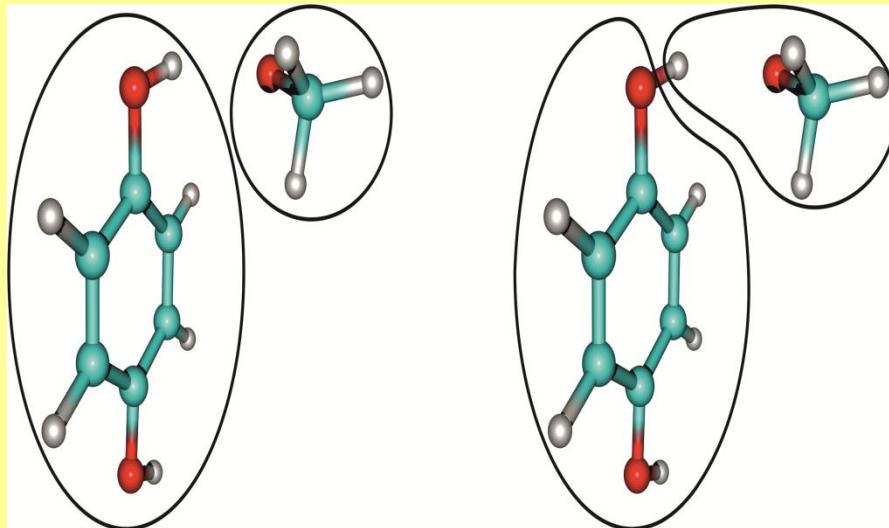
RAZLIČITI REAKCIIONI PUTEVI (1)



RAZLIČITI REAKCIIONI PUTEVI (2)

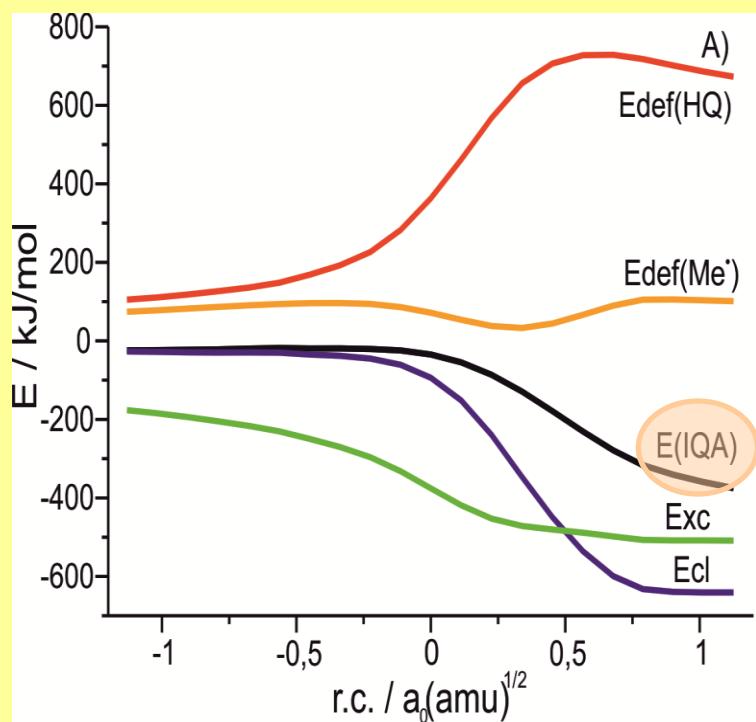


NASTANAK I KIDANJE HEM. VEZE (1)



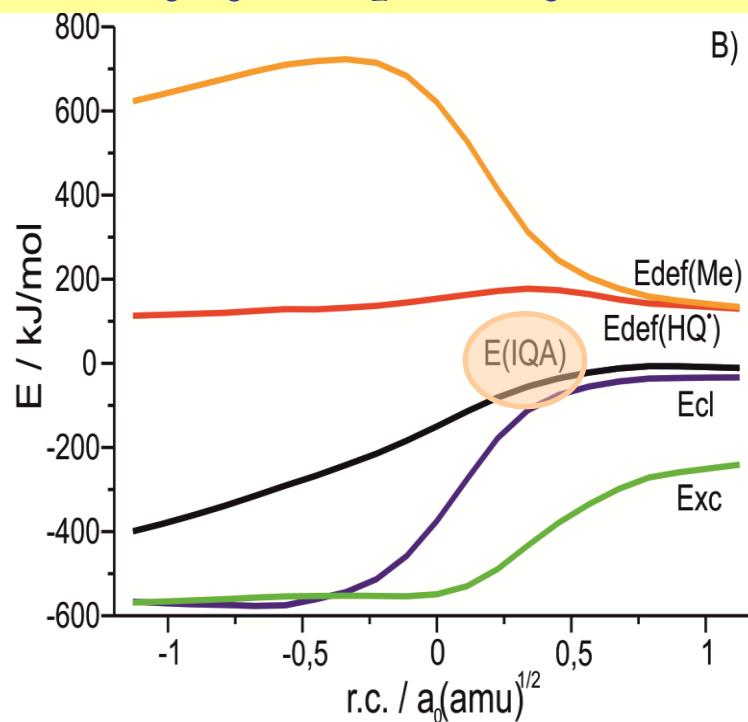
NASTANAK I KIDANJE HEM. VEZE (2)

fragmenti reaktanti:
 $C_6H_4(OH)_2 + CH_3O\cdot$



informacije o vezi
koja nastaje

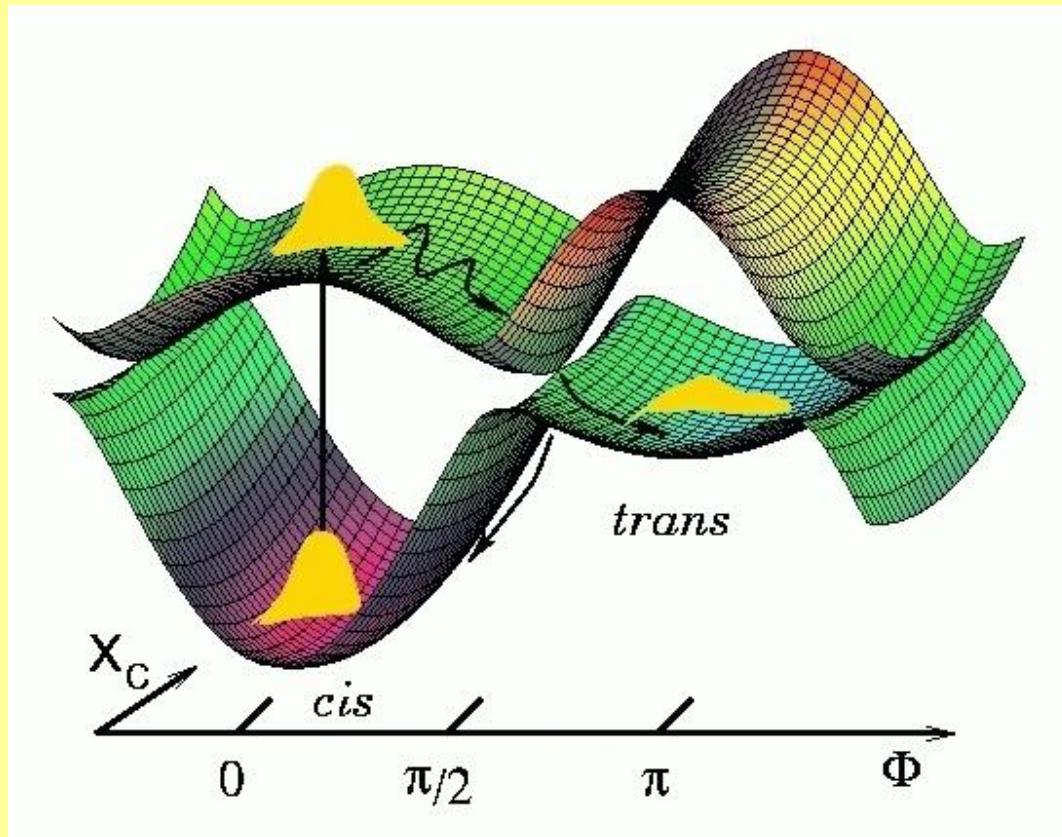
fragmenti proizvodi:
 $C_6H_3(OH)_2\cdot + CH_3OH$



informacije o vezi
koja se kida

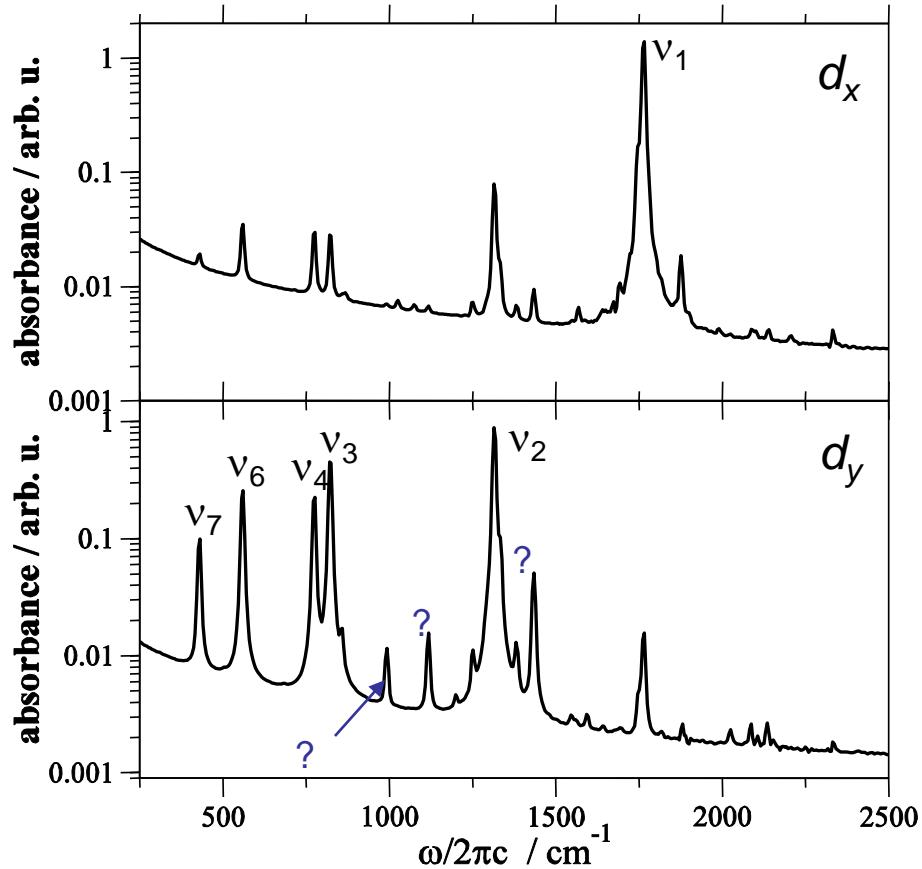
- ultrabrzi procesi
- vremenski nezavisna Šredingerova jednačina
- vremenski zavisna Šredingerova jednačina

ANALIZA DINAMIKE SISTEMA



G. Stock, W. Domcke, *Conical Intersections in Femtosecond time-resolved spectroscopy of the dynamics at conical intersections* (World Scientific Singapore, 2003)

IC SPEKTRI I ASIGNACIJA TRAKA (1)



M. Petković, Chem. Phys. 331 (2007) 438

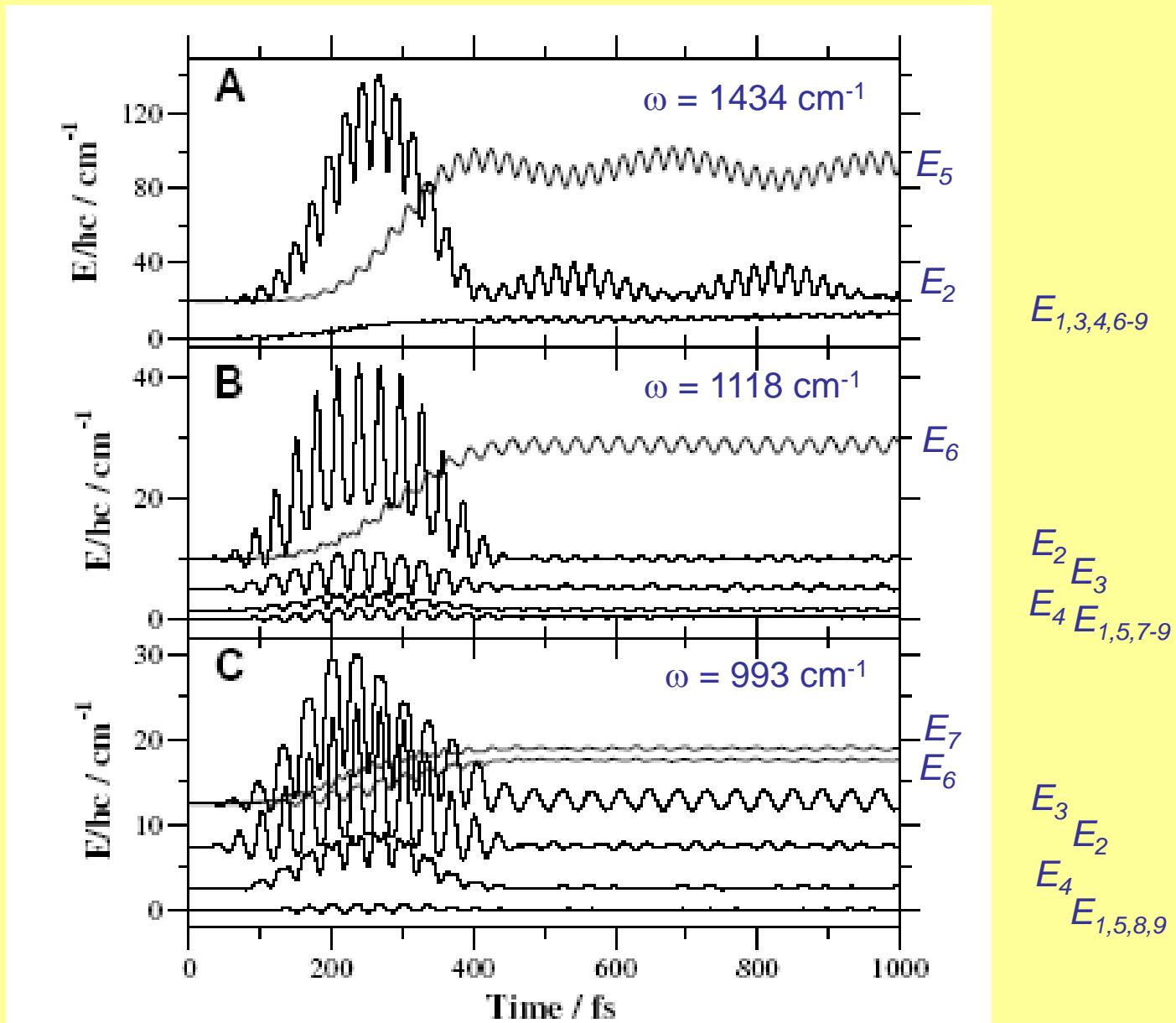
^a Orphal et al. J. Phys. Chem. A 101 (1997) 1062

^b Miller et al. Spec. Acta 23A (1967) 223

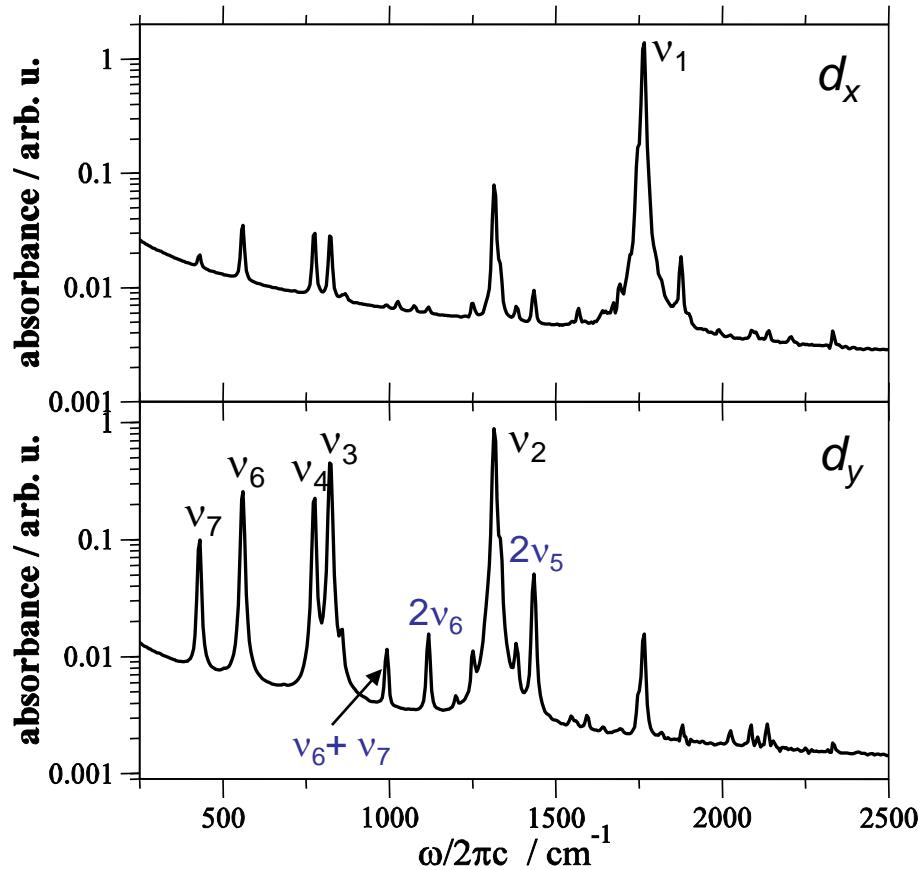
stepen slobode	frekvencija / cm^{-1}		
	harm.	anh.	eksp. ^a
v_1	1800	1766	1737
v_2	1342	1314	1293
v_3	828	820	809
v_4	786	777	780
v_5	731	719	711
v_6	561	561	563
v_7	436	431	434
v_8	250	254	273
v_9	135	158	122

prelaz	frekvencija / cm^{-1}	
	izr.	eksp. ^b
?	1434	1424
?	1118	1119
?	993	988

IC SPEKTRI I ASIGNACIJA TRAKA (2)



IC SPEKTRI I ASIGNACIJA TRAKA (3)



M. Petković, Chem. Phys. 331 (2007) 438

^a Orphal et al. J. Phys. Chem. A 101 (1997) 1062

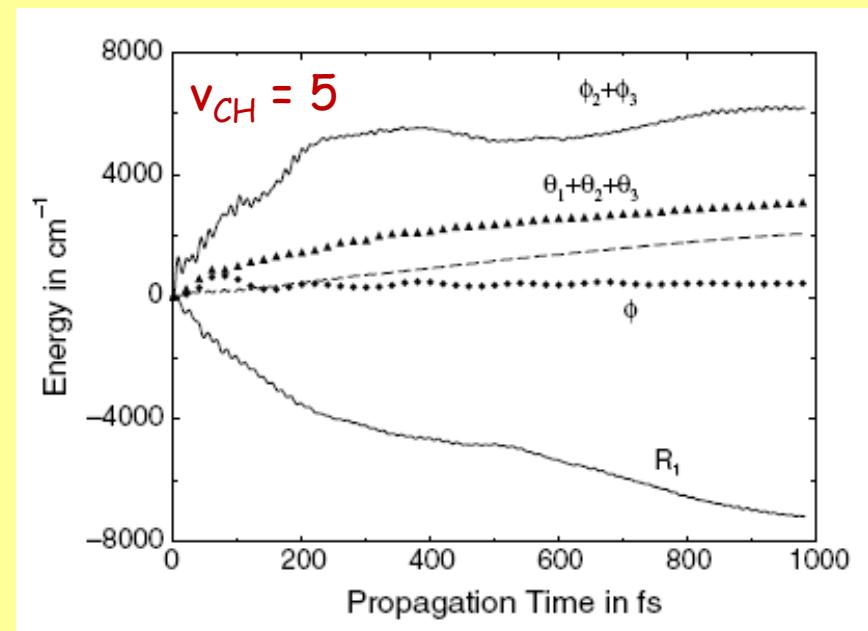
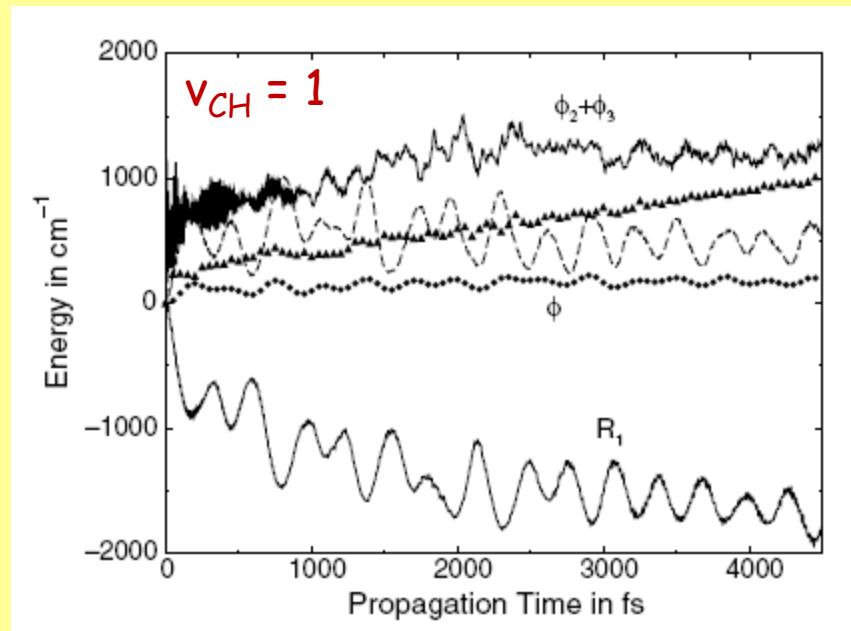
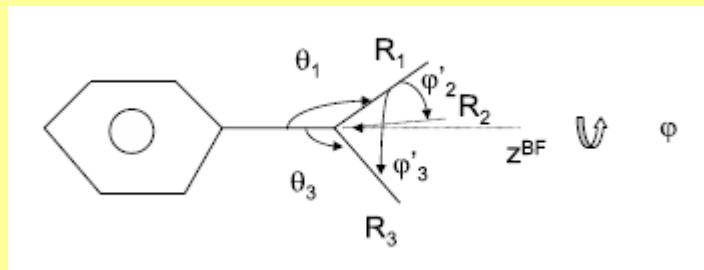
^b Miller et al. Spec. Acta 23A (1967) 223

stepen slobode	frekvencija / cm^{-1}		
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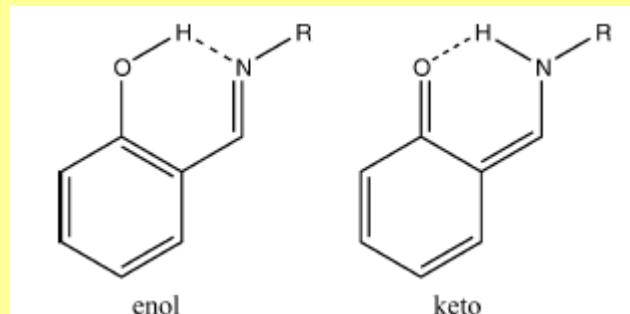
prelaz	frekvencija / cm^{-1}	
	izr.	eksp. ^b
$2v_5$	1434	1424
$2v_6$	1118	1119
$v_6 + v_7$	993	988

INTRAMOLEKULSKA PRERASPODELA VIBRACIONE ENERGIJE

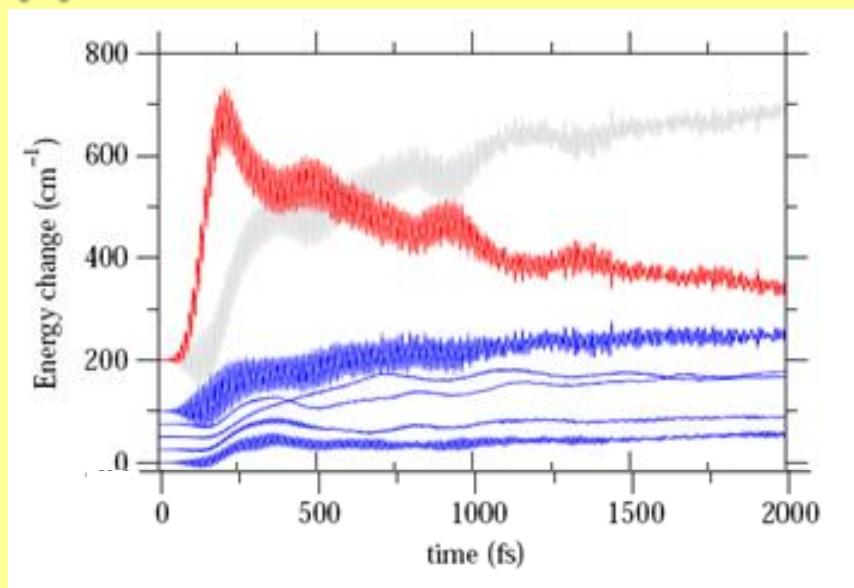
IVR - intramolecular vibrational energy redistribution



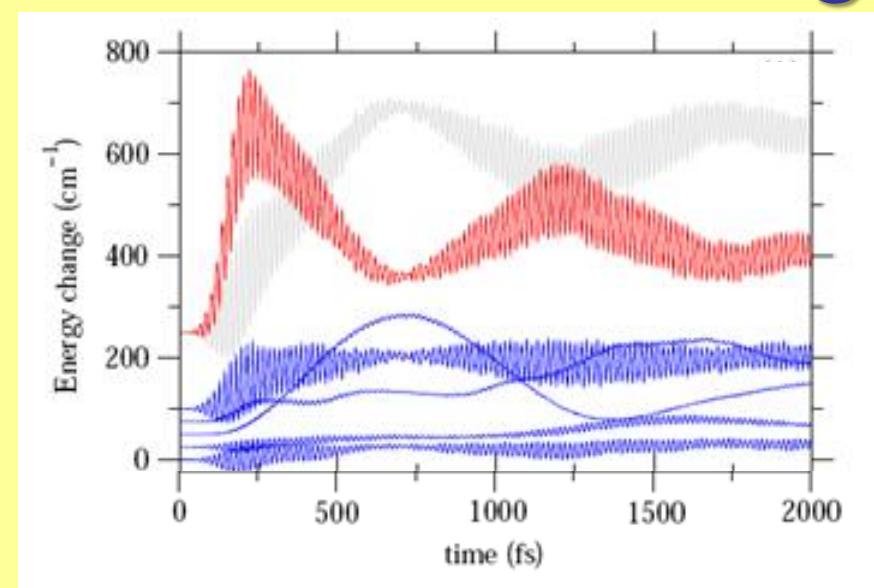
KONTROLA HEMIJSKE REAKCIJE I INTRAMOLEKULSKA PRERASPODELA VIBRACIONE ENERGIJE



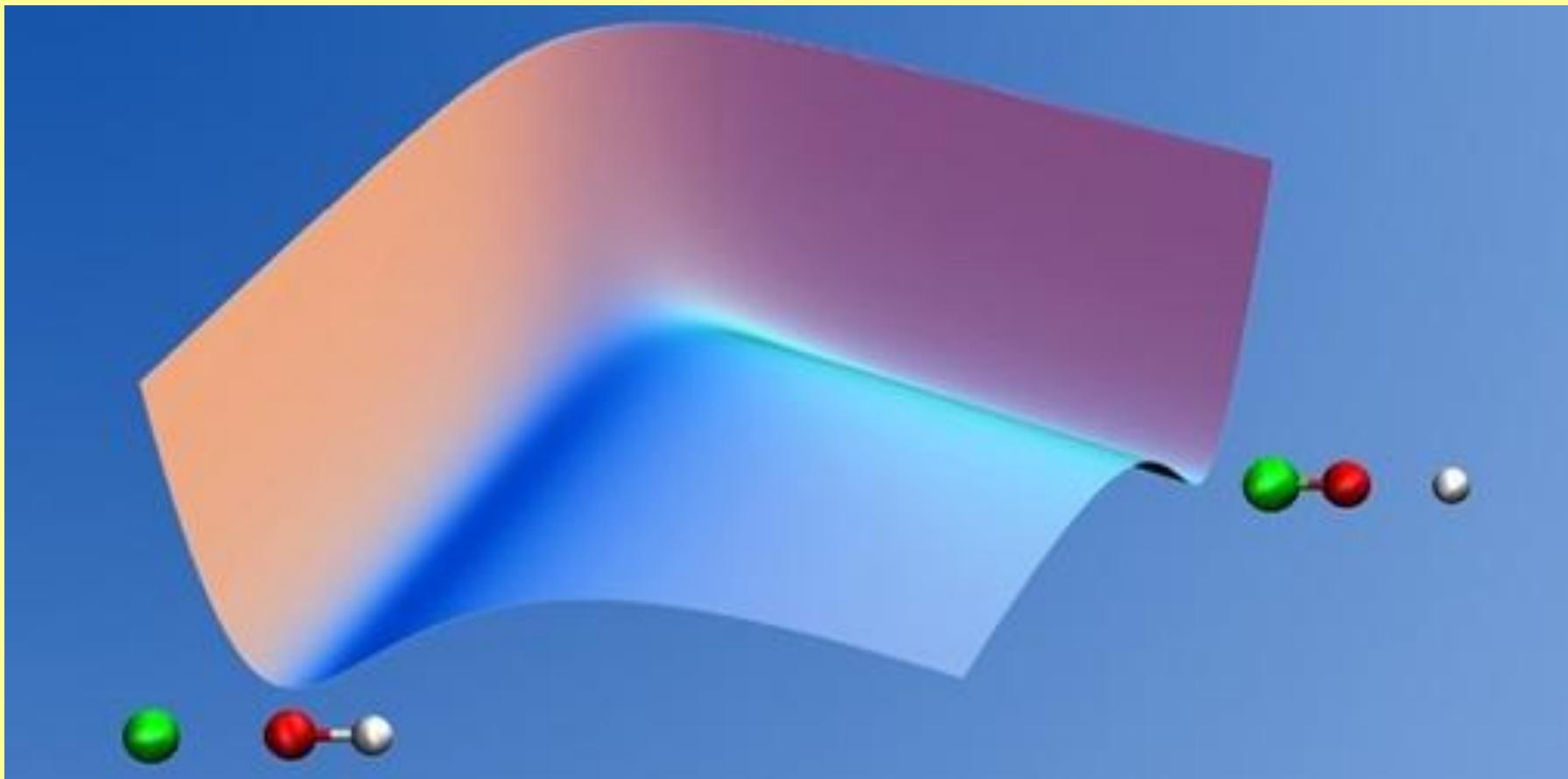
H

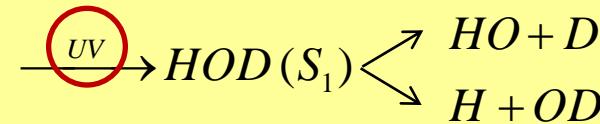
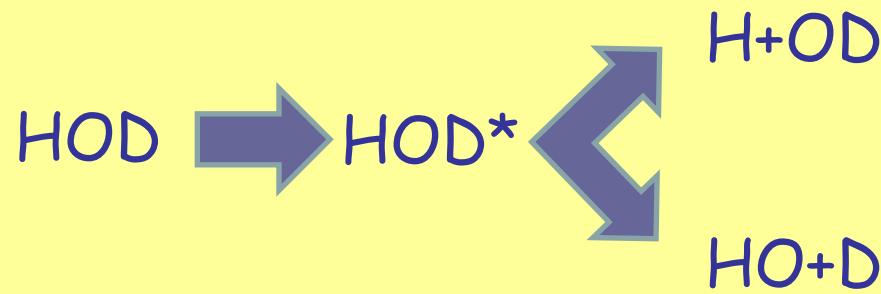


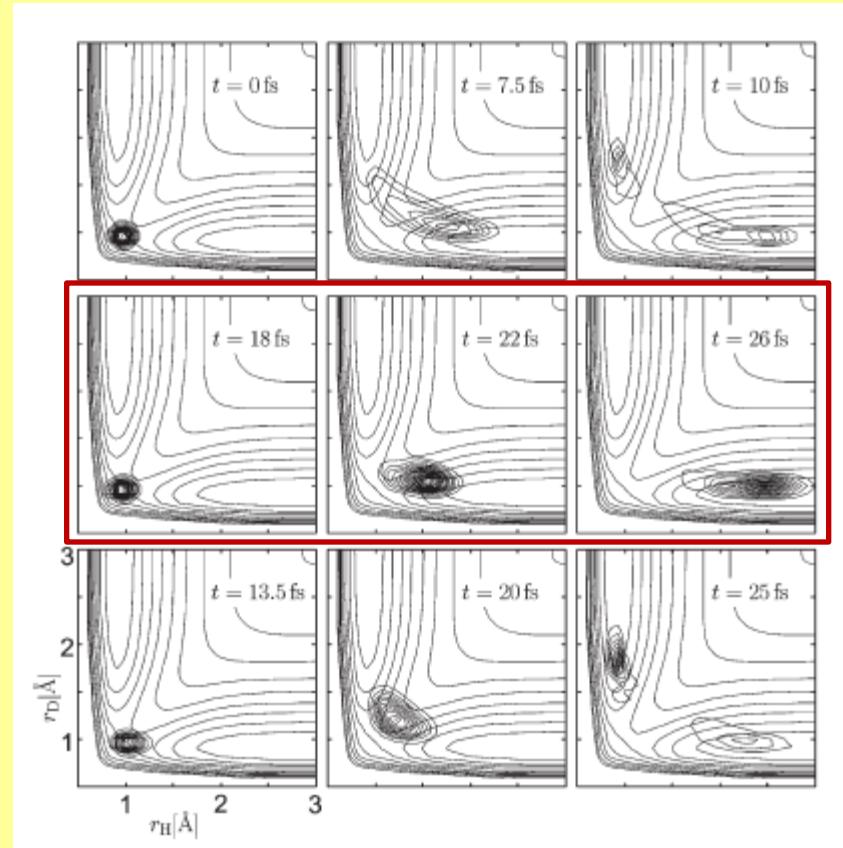
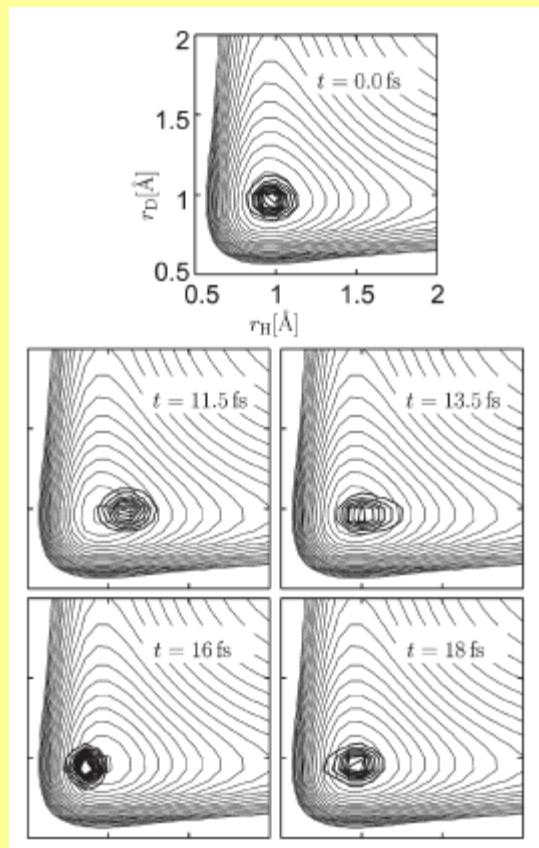
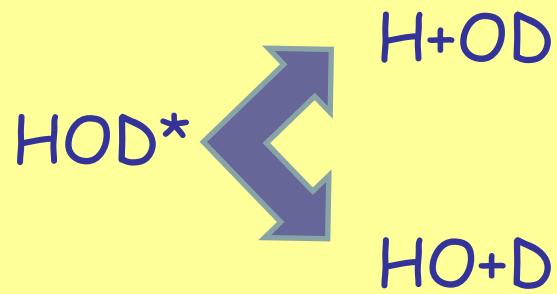
D



KONTROLA HEMIJSKE REAKCIJE

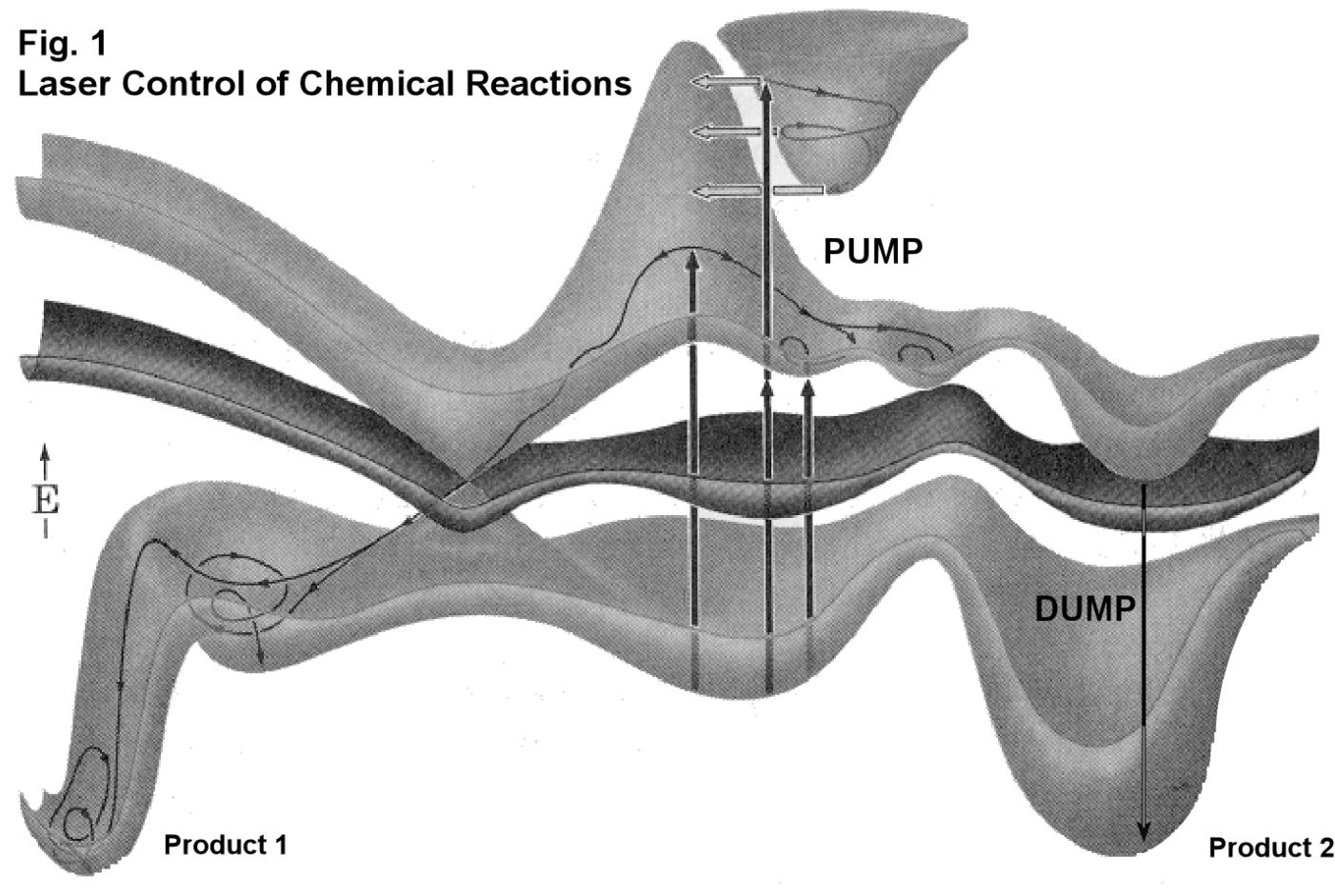






KONTROLA HEMIJSKE REAKCIJE

Fig. 1
Laser Control of Chemical Reactions



J. Michl and V. Bonačić-Koutecky, Electronic aspects of organic Photochemistry,
John Wiley & Sons: New York, 1990

REZIME

- ultrabrzi procesi (atosekundna - nanosekundna skala)
- vremenski nezavisna Šredingerova jednačina
(analiza svojstava sistema duž puta minimalne energije, asignacija spektralnih traka)
- vremenski zavisna Šredingerova jednačina
(vremenska evolucija sistema, kontrola hemijske reakcije)