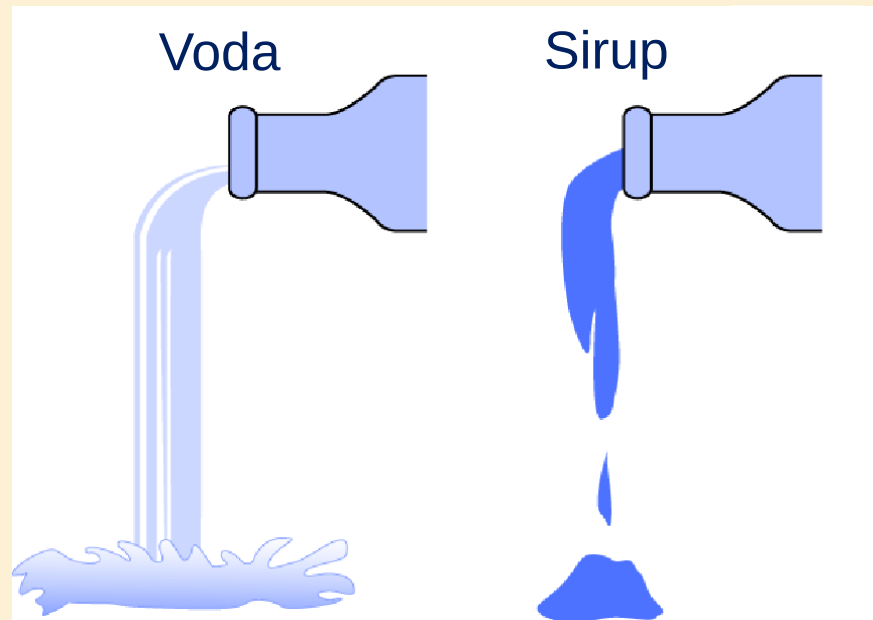


VISKOZNOST TEČNOSTI

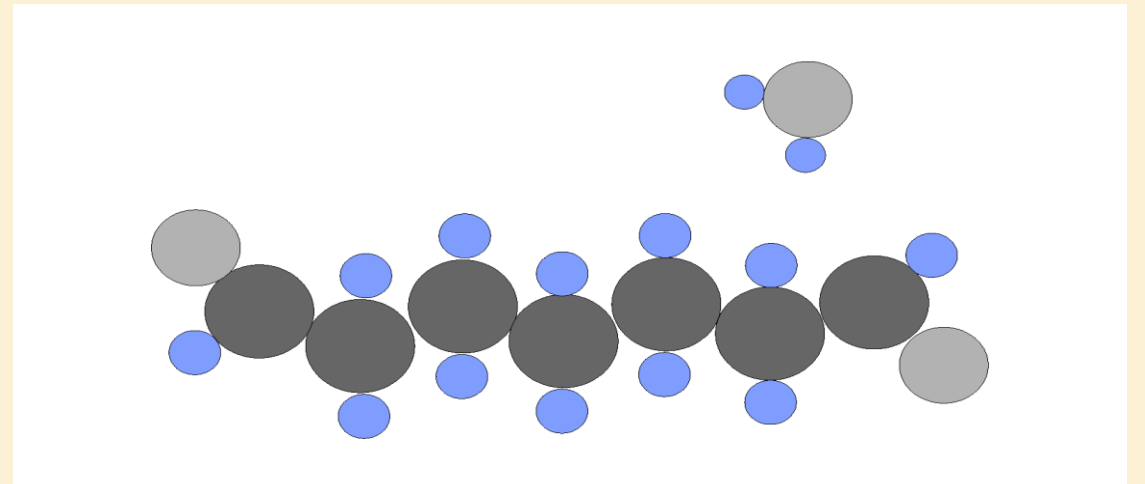
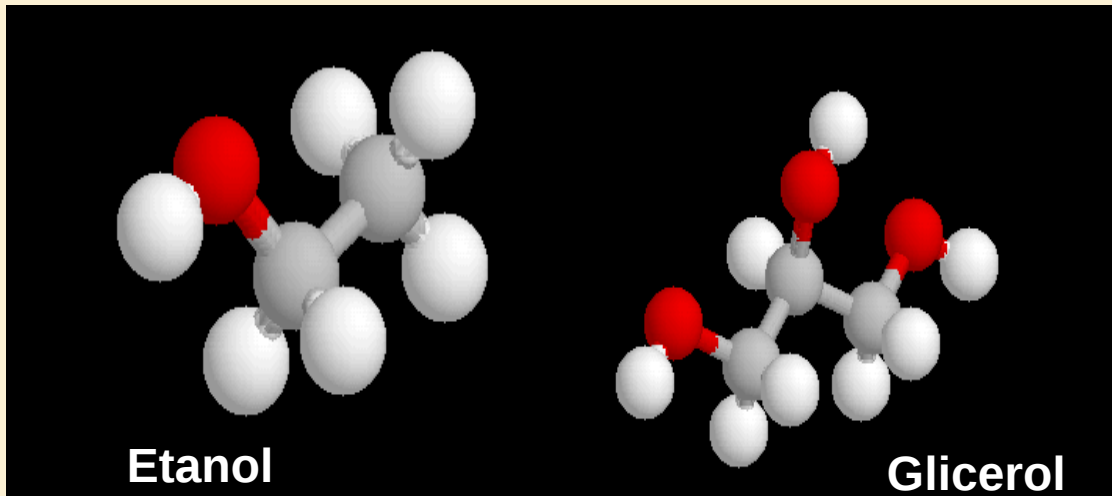
Viskoznost tečnosti

Viskoznost: otpor kojim se pojedini slojevi tečnosti suprotstavljaju kretanju jednih u odnosu na druge (vrsta **unutrašnjeg trenja** koja dovodi do protoka fluida konstantnom brzinom).



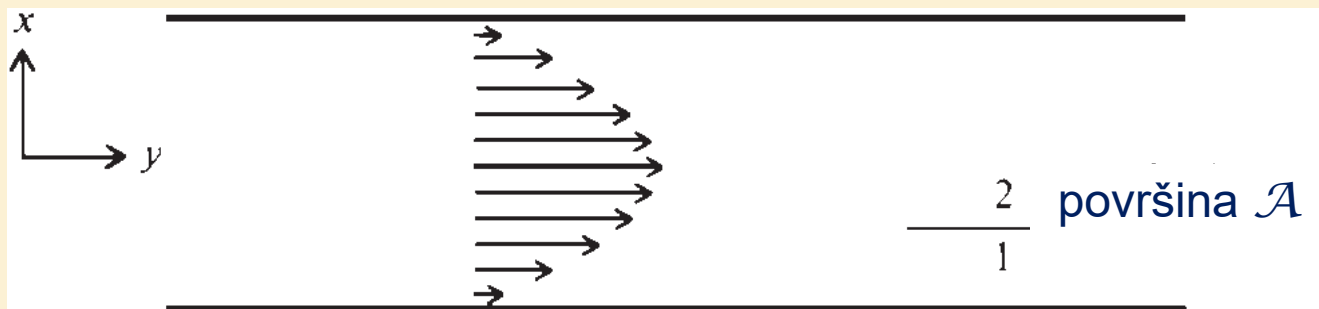
Uzroci otpora proticanju tečnosti

- međumolekulske interakcije
- oblik i veličina molekula



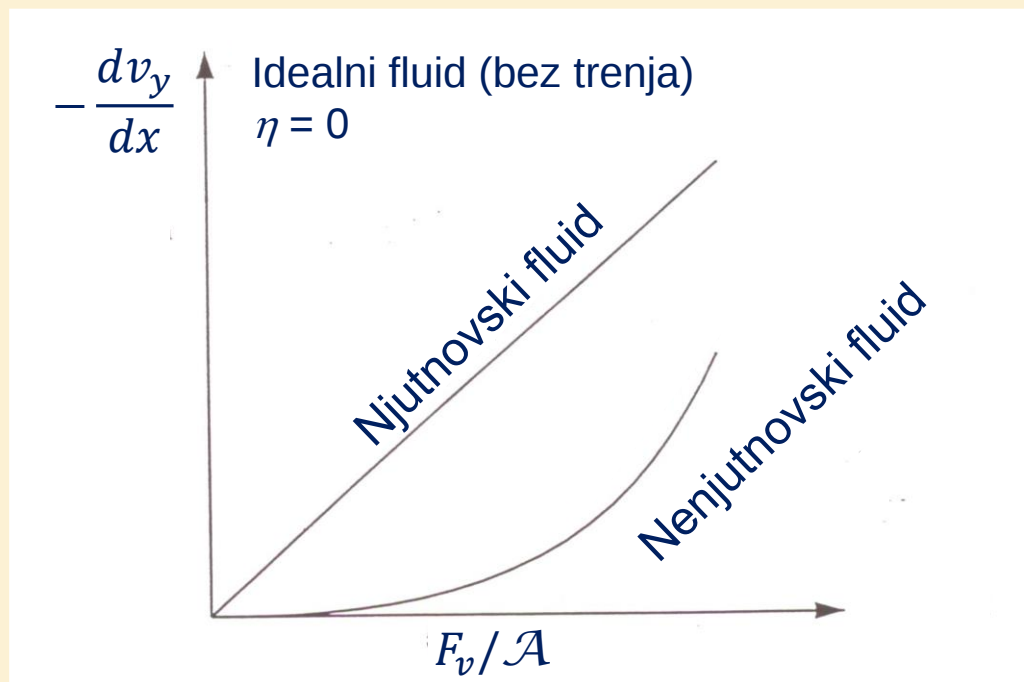
Model

fluid protiče između dve paralelne ploče



Njutnov zakon viskoznosti:

$$F_v = -\eta \mathcal{A} \frac{dv_y}{dx}$$



$$\eta = -\frac{F_v/\mathcal{A}}{dv_y/dx}$$

η : dinamički koeficijent viskoznosti
jedinica je poaz: **1 P = 0,1 Pa s**

Dinamički i kinematički koeficijent viskoznost

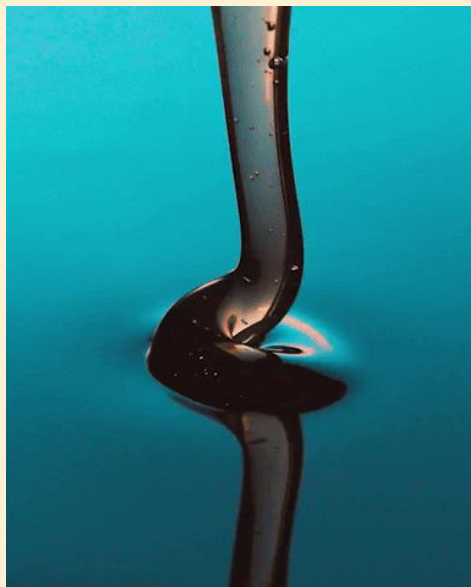
koeficijent viskoznosti	definicija	jedinica
dinamički	$\eta = - \frac{F_v / \mathcal{A}}{dv_y / dx}$	Poaz: 1 P = 0,1 Pa s
kinematički	$\nu = \frac{\eta}{\rho}$	Stoks: 1 St = 10 ⁻⁴ m ² s ⁻¹

Dinamički i kinematički koeficijent viskoznost

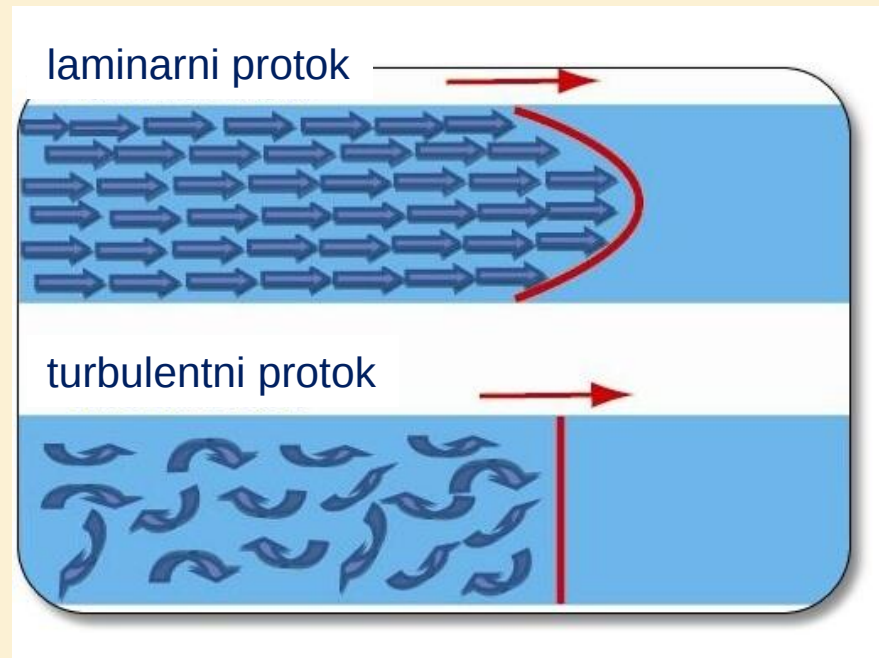
	temperatura / °C	dinam. koef. viskoznost / Pa s	gustina / kg/m ³	kinem. koef. viskoznosti / m ² /s
vazduh	0	$17,09 \cdot 10^{-6}$	1,293	$13,22 \cdot 10^{-6}$
	20	$18,08 \cdot 10^{-6}$	1,205	$15,00 \cdot 10^{-6}$
	40	$19,04 \cdot 10^{-6}$	1,128	$16,88 \cdot 10^{-6}$
sveža voda	0	$1,787 \cdot 10^{-3}$	$1,000 \cdot 10^3$	$1,787 \cdot 10^{-6}$
	20	$1,002 \cdot 10^{-3}$	$0,998 \cdot 10^3$	$1,004 \cdot 10^{-6}$
	40	$0,653 \cdot 10^{-3}$	$0,992 \cdot 10^3$	$0,658 \cdot 10^{-6}$
morska voda	20	$1,072 \cdot 10^{-3}$	$1,024 \cdot 10^3$	$1,047 \cdot 10^{-6}$
aceton	20	$0,326 \cdot 10^{-3}$	$0,792 \cdot 10^3$	$0,412 \cdot 10^{-6}$
glicerin	20	$1,490 \cdot 10^{-3}$	$1,261 \cdot 10^3$	$1,182 \cdot 10^{-3}$
živa	20	$1,554 \cdot 10^{-3}$	$13,546 \cdot 10^3$	$0,115 \cdot 10^{-6}$

Fluidnost

koeficijent fluidnosti: $\phi = \frac{1}{\eta}$
(pokazuje lakoću kojom tečnost teče)



Vrste protoka

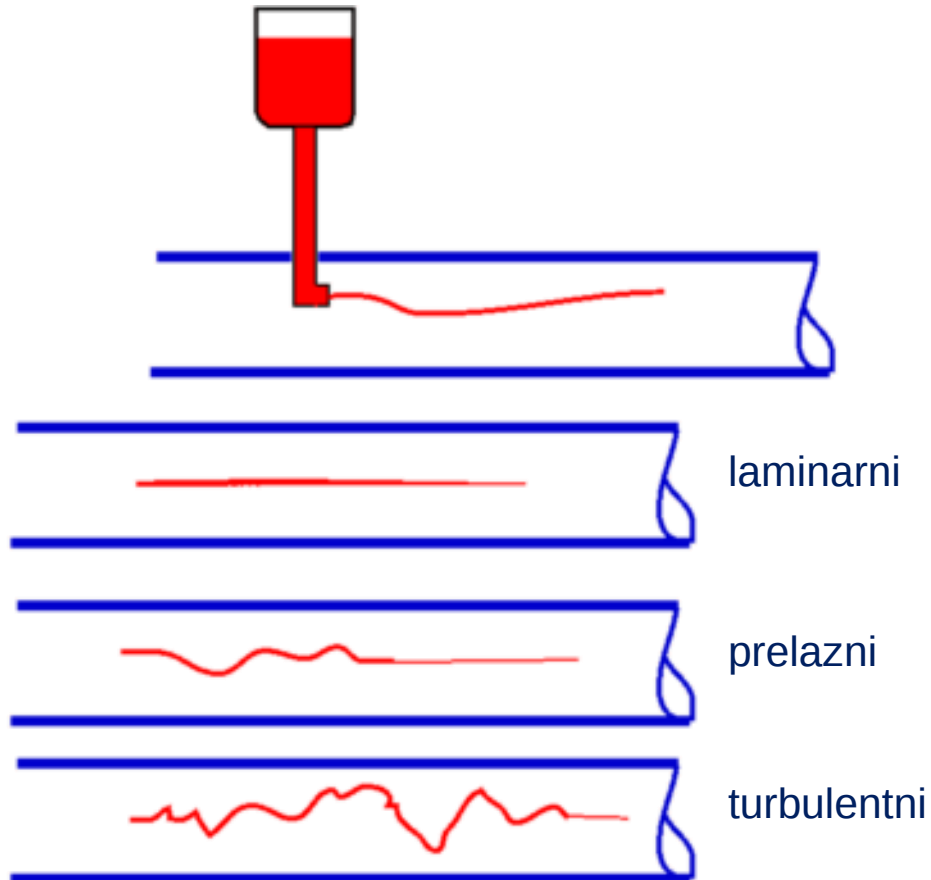


laminarni protok



turbulentni protok

Rejnoldsov eksperiment



$$Re = \frac{\rho v d}{\eta}$$

ρ – gustina fluida

v – srednja brzina kretanja fluida kroz cev

d – prečnik cevi

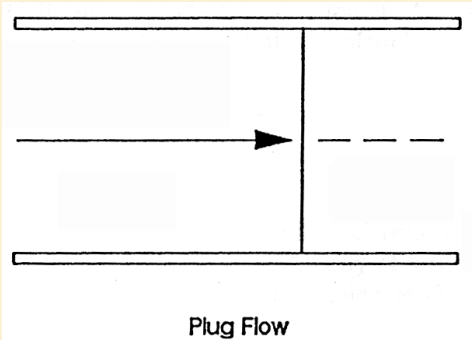
η – koeficijent viskoznosti fluida

Tipovi protoka fluida

$$R_e = \frac{\rho v d}{\eta}$$

Idealni protok

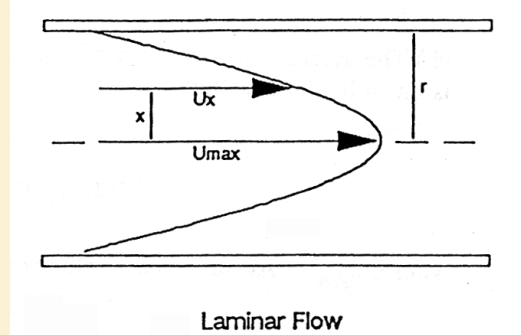
(R_e = beskonačno)



Ovaj tip protoka se ne javlja u praksi.

Laminarni protok

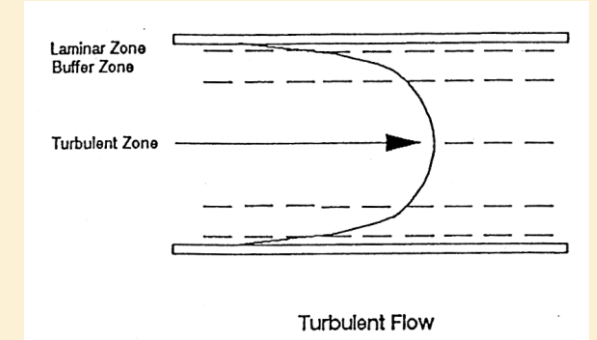
($R_e < 2000$)



Uobičajeni tip protoka pri sporom proticanju.

Turbulentni protok

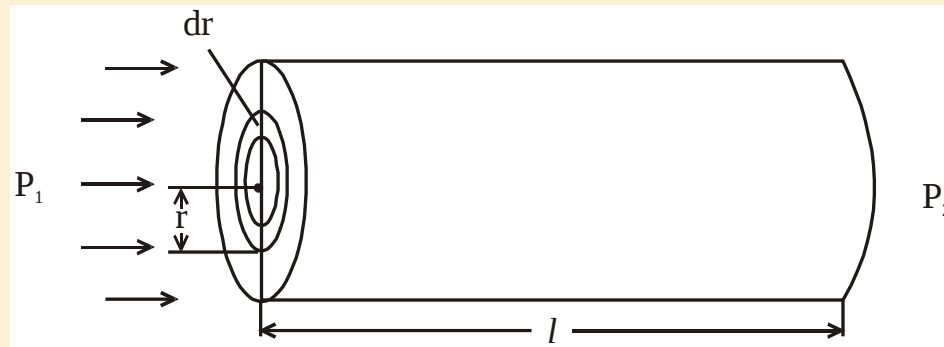
($R_e > 4000$)



Česti tip protoka pri brzom proticanju.

Poazejev zakon

Posmatra se stacionarno proticanje nestišljivog fluida kroz cev pod dejstvom konstantne razlike pritiska.



$$F^r = \Delta P \cdot \pi r^2$$

$$F_v^r = -\eta \frac{dv}{dr} 2\pi r l$$

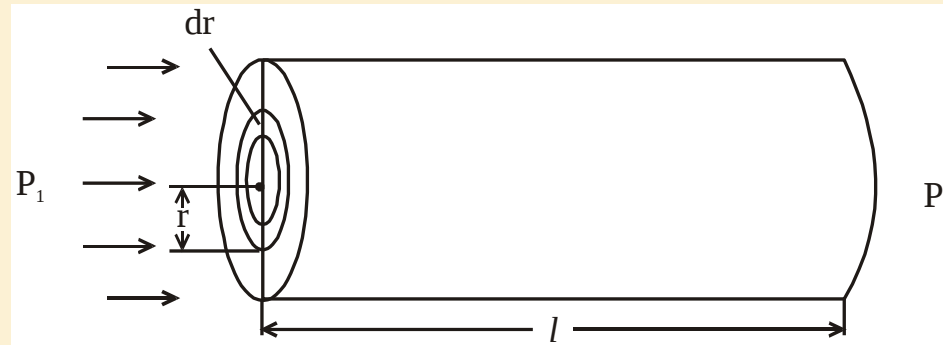
$$-\eta \frac{dv}{dr} \cdot 2\pi r l = \pi r^2 \Delta P$$

$$dv = -\frac{\Delta P}{2\eta l} r dr$$

$$\int_v^0 dv = -\frac{\Delta P}{2\eta l} \int_r^R r dr$$

$$v = \frac{\Delta P}{4\eta l} (R^2 - r^2)$$

Poazejev zakon

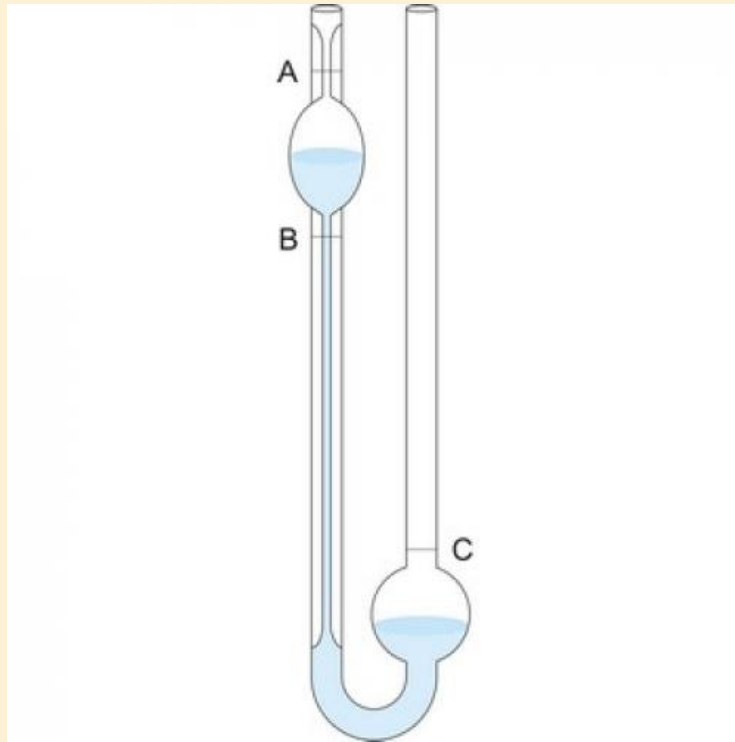


$$dV = vt2\pi r dr = \frac{\Delta P}{2\eta l} \pi (R^2 r - r^3) dr$$

$$V = \frac{\Delta P \pi t}{2\eta l} \int_0^R (R^2 r - r^3) dr$$

$$\boxed{V = \frac{\Delta P \pi R^4 t}{8\eta l}}$$

Merenje koeficijenta viskoznosti: Ostvaldov viskozimetar



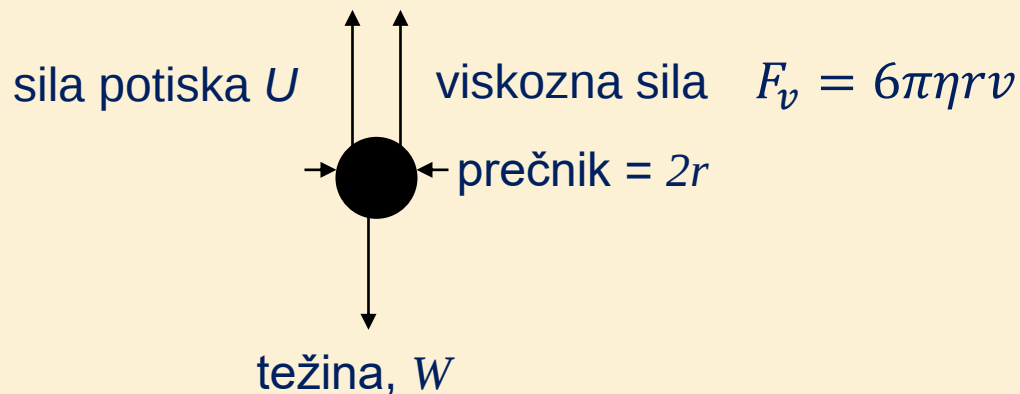
$$\eta = \frac{\Delta P \cdot \pi R^4 t}{8Vl} = \frac{\rho g h \cdot \pi R^4 t}{8Vl}$$

$$\frac{\eta}{\eta_0} = \frac{t}{t_0} \frac{\rho}{\rho_0}$$

Stoksov zakon

$$U = m_l g = \frac{4}{3} \pi r^3 \rho_l g$$

$$W = m_s g = \frac{4}{3} \pi r^3 \rho_s g$$

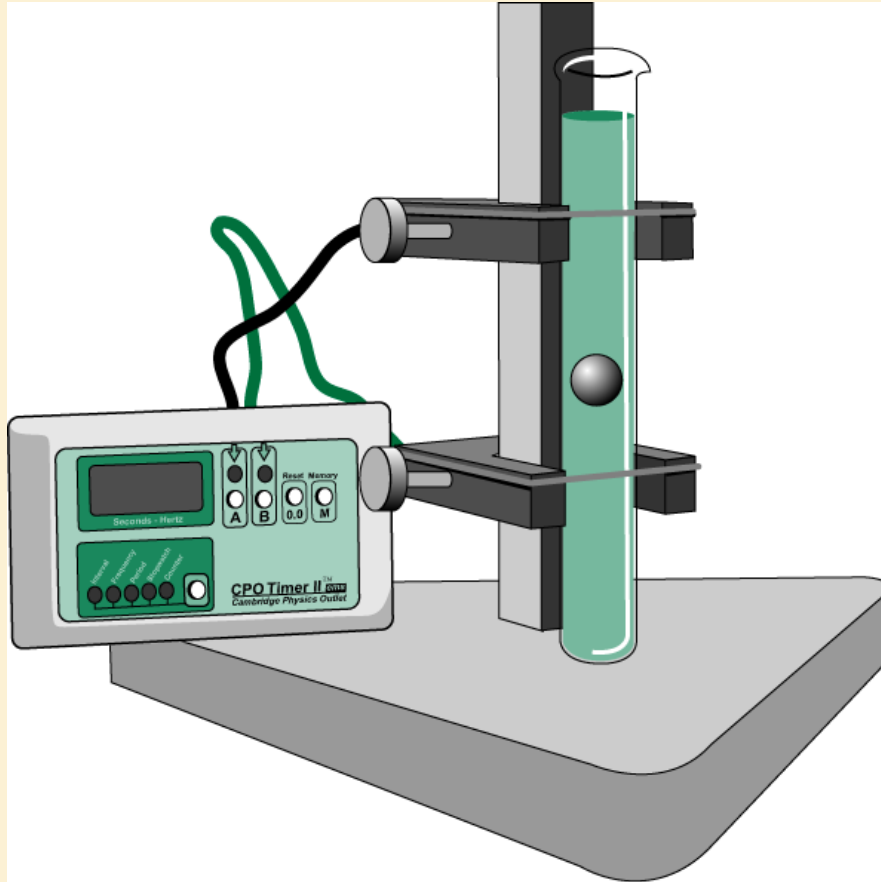


U stanju ravnoteže nema ubrzanja: $U - W + F = 0$

$$\frac{4}{3} \pi r^3 (\rho_l - \rho_s) g + 6\pi\eta r \frac{l}{t} = 0$$

$$\boxed{\eta = \frac{2}{9l} r^2 g (\rho_s - \rho_l) t}$$

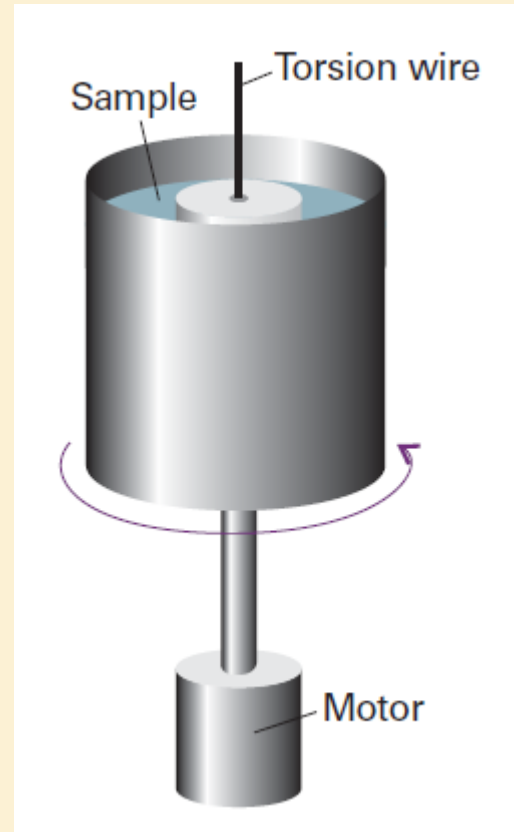
Merenje koeficijenta viskoznosti: Stoksov zakon



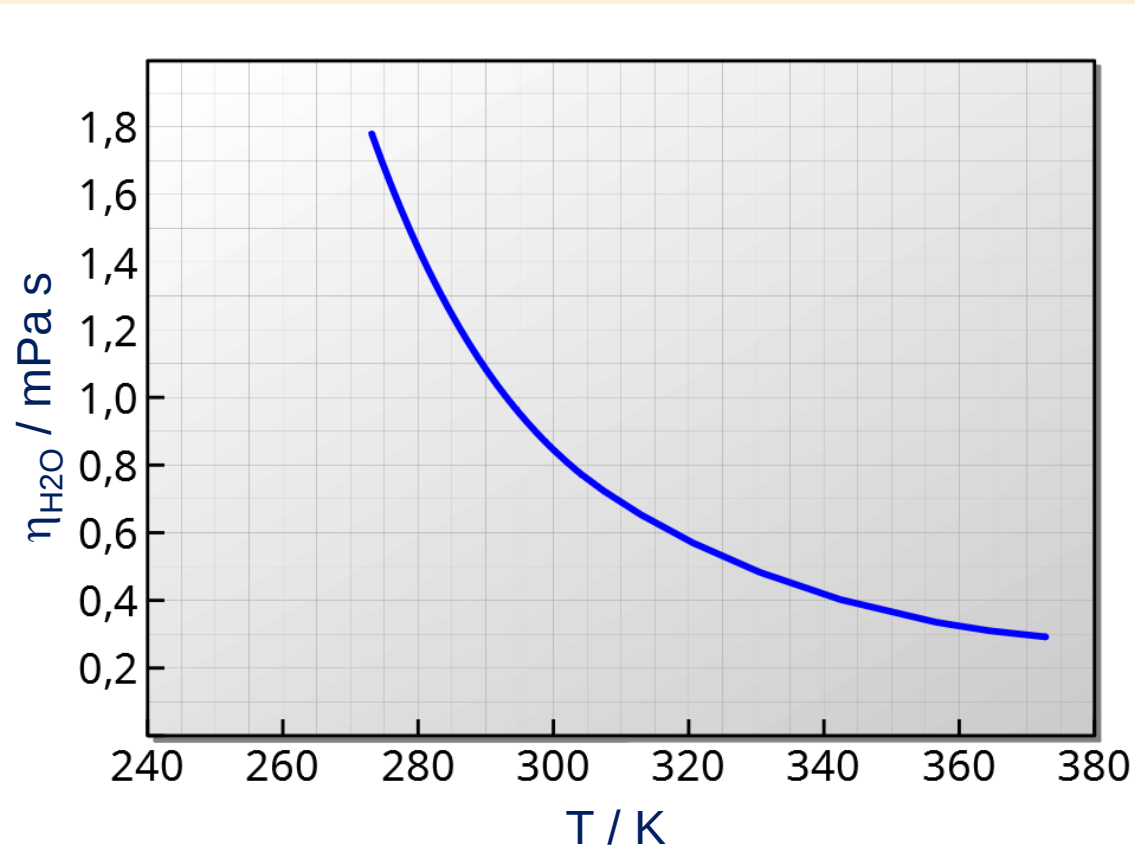
$$\eta = \frac{2t}{9l} r^2 g (\rho - \rho')$$

$$\frac{\eta_1}{\eta_2} = \frac{(\rho - \rho'_1)t_1}{(\rho - \rho'_2)t_2}$$

Merenje koeficijenta viskoznosti: rotacioni viskozimetar



Viskoznost i temperatura



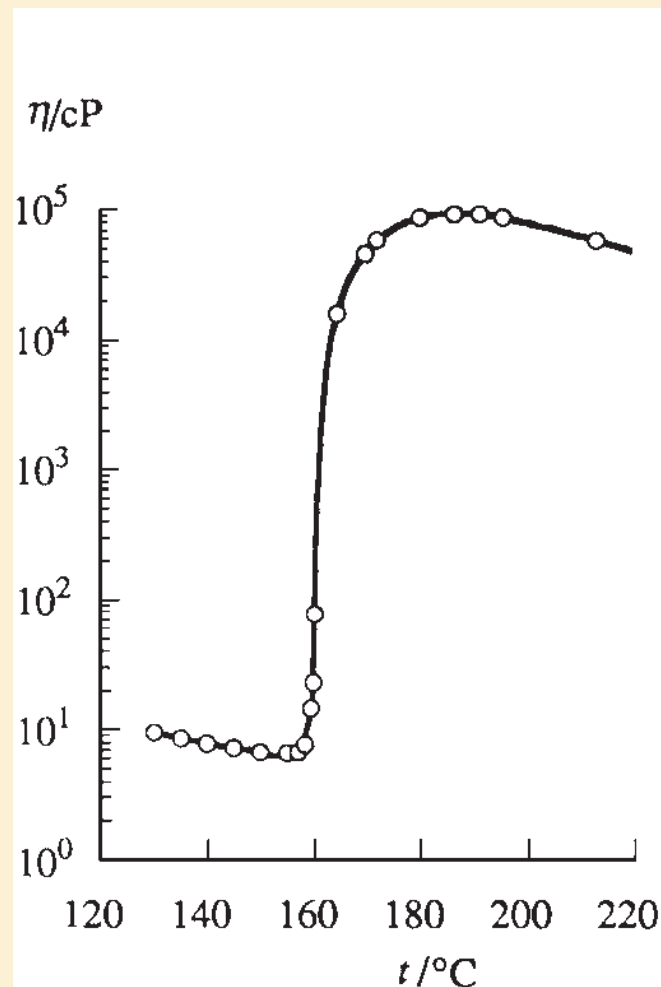
Arenijus i Gucman: $\eta = A \exp\left(\frac{B}{RT}\right)$

Andrade: $\eta v_{sp}^{1/2} = A \exp\left(\frac{B}{RT v_{sp}}\right)$

Bačinski: $\eta = \frac{c}{v_{sp} - b}$

$v_{sp} - b \approx$ zapremina šupljina

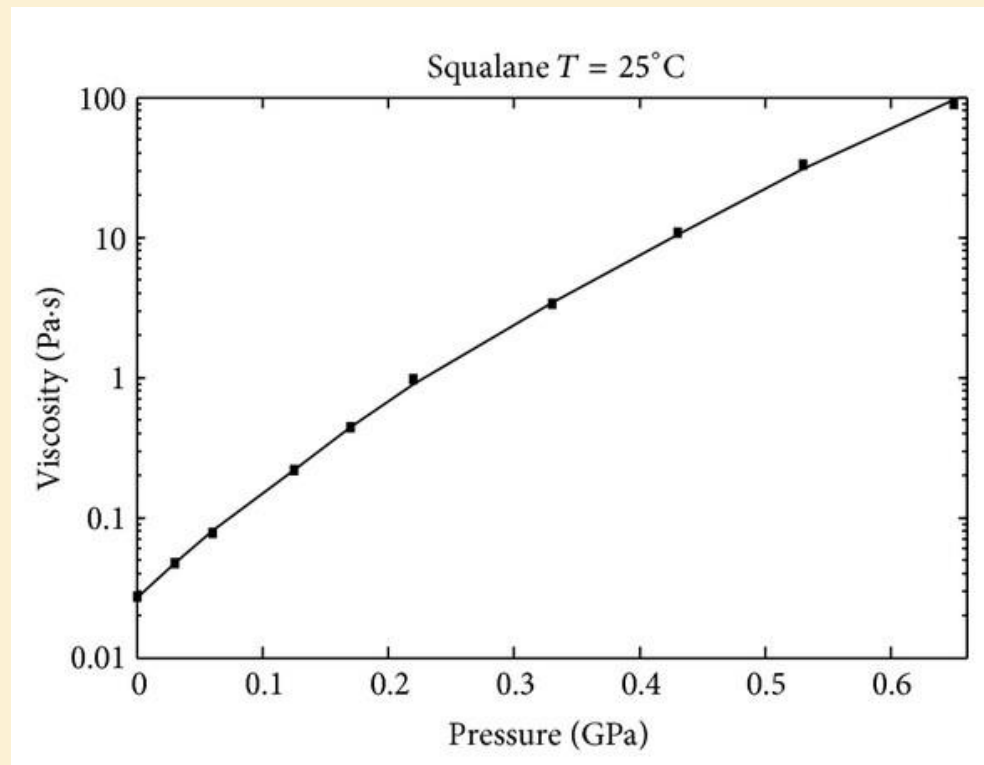
Anomalous behavior of liquid sulfur



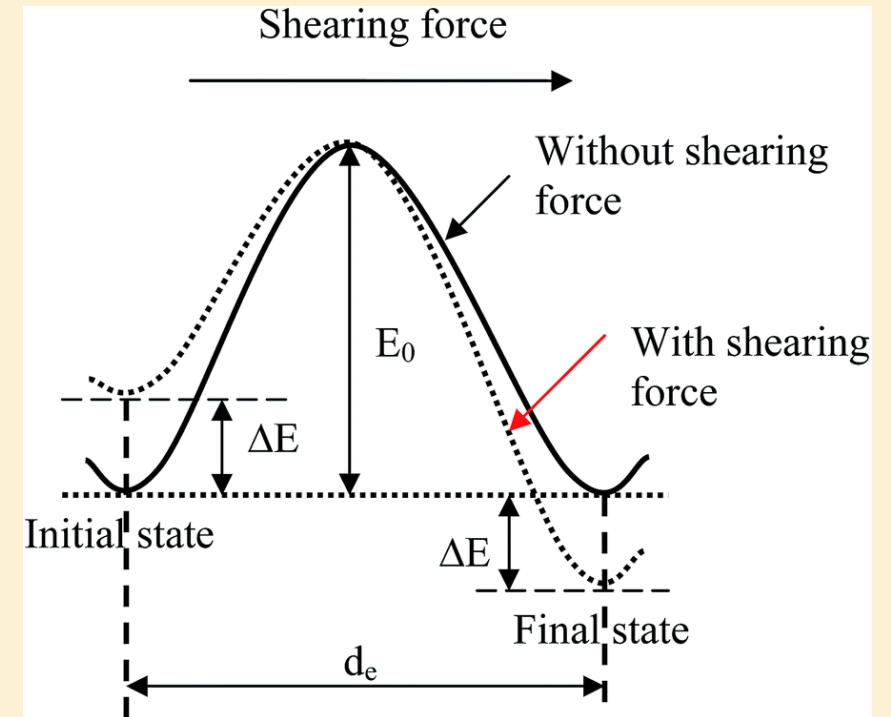
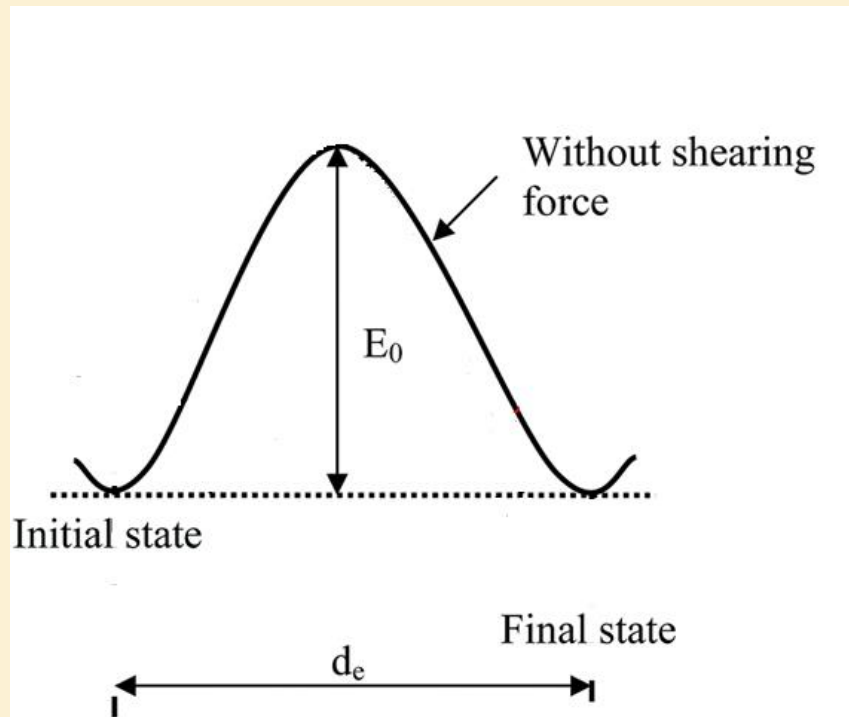
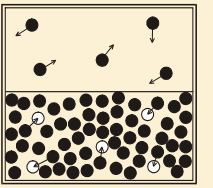
Viskoznost i pritisak

$$\eta = \eta_0 \exp\left(\frac{P\Delta V_h}{RT}\right)$$

ΔV_h - zapremina šupljina po molu tečnosti



Ajringov model značajnih struktura



Viskoznost binarnih sistema

Relativna viskoznost:

$$\eta_r = \frac{\eta}{\eta_0}$$

Specifična viskoznost:

$$\eta_{sp} = \frac{\eta - \eta_0}{\eta_0} = \frac{\eta}{\eta_0} - 1 = \eta_r - 1$$

Redukovana viskoznost:

$$\eta_{red} = \frac{\eta_{sp}}{c}$$

Unutrašnja viskoznost:

$$[\eta] = \lim_{c \rightarrow 0} \frac{\eta_{sp}}{c}$$

Primer: polimeri

Mark-Houvinkova relacija:

$$[\eta] = K \bar{M}^a$$

$[\eta]$ - unutrašnja viskoznost

K, a - konstante za posmatrani sistem polimer – tečnost

$\bar{M}$ - srednja molarna masa polimera